Macomb Regional Science Olympiad Water Quality October 23, 2005

1. The two major purposes of sampling in the water environment are:

2. Why is coliform bacteria used as an indicator of poor water quality?

d. To decide whether or note you can drink it

a. To establish a baselineb. To determine the pollution

c. To test for acid

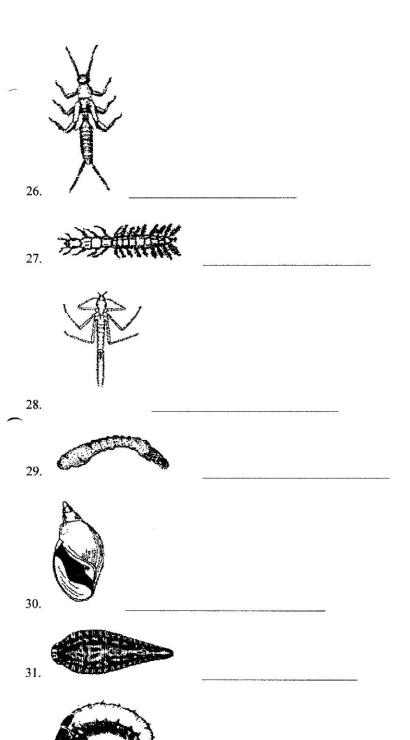
d. All of the above

Part 1 - Multiple-Choice Questions (multiple answers are possible: a & c or a, c, and	(d
---	----

	a. Indicates acid rain
	b. Indicates high O2 levels
	c. Indicates fecal contamination
	d. Indicates high nitrate levels
3. A	water sample with a pH of 2.0 is considered to be
	a. Basic
	b. Acidic
	c. Neutral
	d. Perfect for aquatic life
4. Th	e test for turbidity describes what characteristics of water?
	a. Odor
	b. Mineral concentration of the water
	c. Suspended material in the water
	d. Metal concentration of the water
5. As	secchi disk is used to measure
	a. The dissolved material in the water
	b. Light penetration of a lake or pond
	c. Flow of a stream or river
	d. The depth of silt on the bottom of a lake
Did you kı	now this answer?
6, Th	e alkalinity of a water sample may be defined at the
,	a. Ability of a water sample to evaporate at low temperature
	b. Capacity of a water sample to react with and neutralize acid
	c. Capacity of a water sample to become toxic
	d. Ability of a water sample to carry nutrients to plant
Did y	ou know this answer? _^
7. Th	e quality of drinking water can be influenced by
	a. Gases and aerosols from the atmosphere
	b. Weathering and erosion of rocks and soils
	c. Human activities

8.	The ac	ddition of additional nitrogen and phosphorus to aquatic systems will	
		Increase algae and decrease O2	
	b.	Increase O2 and decrease algae	
	c.	Increase the number of fish	
	d.	Decrease productivity	
		d you know this answer?	
9.		vo main sources of wetlands water are	
	a.	Streams and ponds	
	b.	Oceans and streams	
	c.	Ground water systems	
	d.	Surface water and groundwater	
	e.	Surface water and runoff	
10.	As the	water quality index (WQI) of a stream increases, the biodiversity	
	a.	Increases	
	b.	Deceases	
	c.	Remains the same	
	d.	Is static	
		d you know what static means?	
11.	Most a	aquatic organisms have a pH range of	
	a.	7.0-9.0	
	b.	6.0-9.0	
	c.	4.0-10.0	
	d.	5.0-9.0	
12. At a pH of 7 the concentration of H and OH are			
	a.	Unbalanced H	
		Inequilibium	
		Balanced	
		Unbalanced OH	
13.		ally, daily pH cycles peak:	
		In the morning	
		In the evening	
		At Noon	
	d.	\mathcal{E}	
14.	The an	mount of oxygen in any body of water depends on	
	a.	The amount of animal species diversity	
	b.	The amount of plant species diversity	
		Water temperature, number of green plants, and sunlight	
	d.	All of the above	

- 15. More oxygen is dissolved in
 - a. Warm water and high pressure (high elevation)
 - b. Cold water and low pressure (low elevation)
 - c. Cold water and high pressure (high elevation
 - d. Cold water and high pressures (high elevation)
- 16. Which holds more dissolved oxygen?
 - a. Water at 20° C
 - b. Water at 10°C
 - c. Water at 5°C
 - d. Water at 25°
- 17. Which of the following is a non-point source of pollution to a river or stream?
 - a. Industrial effluent
 - b. Sewage Effluent
 - c. Livestock grazing
 - d. Mining Effluent
- 18. The pH sampling procedure is to collect the water sample at
 - a. The river bank
 - b. Near the bottom of the river
 - c. Away from the river bank and below the surface
 - d. It does not matter where you collect the sample
- 19. List two human-caused changes in the temperature of a river
- 20. What type of pollution is a change in temperature?
- 21. How would you test for thermal pollution?
 - a. Test in one location in the morning
 - b. Test downstream
 - c. Test upstream near the source of the river
 - d. Test above and below where the suspected source
- 22. True or false, the greater the turbidity, the murkier the water. _____
- 23. Which of the following are causes of turbidity
 - a. Soil erosion
 - b. Waste Discharge
 - c. Urban runoff
 - d. All of the above
- 24. Why does water become warmer as more suspended particles enter a river?
- 25. True or false, DO levels do not vary much according to time, weather, and temperature.



32.

- 1. a&b
- 2. c
- 3. b
- 4. c
- 5. b.
- 6. b
- 7. d
- 8. a
- 9. d. 10. a 11.a
- 12. c
- 13. d
- 14. d
- 15. d
- 16. c
- 17. c
- 18. c
- 19. industries such as power plants, cutting down trees, street runoff
- 20. thermal pollution 21.d
- 22. true
- 23. d
- 24. suspended particles absorb heat from sunlight
- 25. false
- 26. stonefly
- 27. whirlgig larvae
- 28. damselfly nymph
- 29. black fly larvae
- 30. lunged snail
- 31. leech
- 32. caddis fly larvae